Supplemental Instruction – Biology 2300
SI Leader – Philipp Orbe
Session: 1 – Cell Biology Introduction
1. What are the two basic forms of cells? What are some defining characteristics?
2. Bacteria and archaea are cells.
3. Name the region of coiled DNA present in prokaryotic cells.
<ul> <li>Is the DNA in prokaryotes surrounded by a membrane?</li> </ul>
4. T/F: Eukaryotes are the only cells that have a cell wall.
5. Some prokaryotes have surface projections. List some surface projections and their function.
6. Prokaryotes and eukaryotic cells are similar in several different ways, one of which is the
presence of ribosomes. What is the function of ribosomes? In terms of bacteria why is this an important function?
7. What are the four elements that make up 96% of all living matter?

8.	What is the difference between a element and a compound?	
9.	Let's talk about atoms and their three subatomic particles. List them and their defining characteristics.	
10.	Draw out an atom. (Nucleus and orbital space).	
	What helps keep electrons near the nucleus?	
11.	What's the difference between the atomic number and the mass number?	
12.	Electron shells contain electrons and the electrons contained in the outermost shell of an atom are called electrons.	
13.	: An element with a full outer electron shell	
14.	List and define the two types of bonds that interact by sharing or donating/receiving electrons.	

15.	5. What is the name of the structure that is formed when two or more atoms are held together by		
	covalent bonds?		
16.	T/F: Protons and electrons determine an atom's electronegativity.		
17.	What is the difference between nonpolar and polar covalent bonds?		
18.	A weak attraction of hydrogen to any negative particle is a hydrogen bond. What are some		
	examples?		
19.	In terms of ionic bonding, an ion with a negative charged is an, and a positively		
	charged ion is a		
20.	T/F: The products and reactants of a chemical reaction are present in equal amounts.		
21.	What is the defining characteristic of organic compounds?		
22			
22.	How many bonds can carbon form? Why?		
23.	Functional groups we will cover include Amino, Phosphate, and Methyl. Draw each.		

24. What are the 4 classes of macromolecules? What are the building blocks of each?			
	•		
	•		
	•		
	•		
25.	Building blocks are called and are linked together to form through		
	which remove water. The breakdown of these bonds is called, the		
	addition of water.		
26.	We will cover 4 main carbohydrates in this class, list them and their purpose.		
27.	T/F: Lipids repel water and are said to be hydrophobic.		
28.	Other than energy storage, the other important function of lipids is construction of the		
29.	A large lipid made of two smaller molecules (glycerol and fatty acids) is called a What are		
	some of the characteristics?		
30.	Fatty acids can contain no/one/or more than on double bond. If a fatty acid has a double bond it		
	is considered a A fatty acid with no double bonds has the maximum number of		
	hydrogens and is called a . Draw each kind of fatty acid.		

31.	Which kind of lipid is the major	component of all cell membranes?
32.	In terms of the membrane phos	pholipids cluster together and form a membrane with two layers
	called a The	heads are in contact with the environment and the
	internal part of the cell. The	tails cluster together in the center and face each other.
33.	·	f the cell membrane (used to increase membrane It is also a starting material for making steroids (including sex rogen).